

Substitute for Form 1449/PTO

INFORMATION DISCLOSURE STATEMENT BY APPLICANT

(use as many sheets as necessary)

Sheet	1	of 2	Attorney Docket Number BSTZ Docket No.	Complete if Known
				Application Number 09/891,730
				Filing Date June 25, 2001
				First Named Inventor: Steven Verhaverbeke
				Art Unit 1765
				Examiner Name Umez-Eronini, L.

U.S. PATENT DOCUMENTS

Examiner Initials* CITE	Cite No. ¹	Document Number		Publication Date MM-DD-YYYY	Name of Patentee or Applicant of Cited Document	Pages, Columns, Lines, Where Relevant Passages or Relevant Figures Appear
		Number-Kind Code ² (If known)				
US- 3,045,702	07-24-1962	Nakata				
US- 3,291,347	12-13-1966	Blades				
US- 4,243,071	01-06-1981	Shackelford				
US- 4,554,050	11-19-1985	Minford et al.				
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FOREIGN PATENT DOCUMENTS

Examiner Initials*	Cite No. ¹	Foreign Patent Document	Publication Date MM-DD-YYYY	Name of Patentee or Applicant of Cited Document	Pages, Columns, Lines, Where Relevant Passages or Relevant Figures Appear	T ⁶
		Country Code ³ Number ⁴ Kind Code ⁵ (if known)				
ZJME	EP 0 245 667 A1		11-19-1987	Edeleanu Gesellschaft mbH		No

Examiner Signature	Lynette T. Umez-Eronini	Date Considered	9/27/2004
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Sheet

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Examiner Name	Umez-Eronini, L.
Attorney Docket Number	4990 USA/W-C/W-C/JBI
bstz Docket No.	4887.P447

NON PATENT LITERATURE DOCUMENTS

Examiner Initials*	Cite No ¹	Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published	T ²
L.J.N.E.		Chapter 3 – Silicon Wafer Cleaning Procedure, http://www-mtl.mit.edu/CAFE/sop_copy/rca.html , 04/24/2002, 4 pages.	
L.J.N.E.		Patent Abstracts of Japan , Vol. 004, No. 089 (C-016), June 25, 1980 & JP 55 051427 A (Sakaoka Kazuhiko), April 15, 1980, 1 page.	
L.J.N.E.		Silicon VLSI Technology, Fundamentals, Practice and Modeling, By Plummer, Deal and Griffin, IC Manufacturing-Chapter 4, Semiconductor Manufacturing-Clean Rooms, Wafer Cleaning And Gettering- Chapter 4, ©2000 by Prentice Hall, Upper Saddle River, N.J., 16 pages.	
L.J.N.E.		Written Opinion for PCT/US/01/41160 mailed April 2, 2004, 7 pages.	
L.J.N.E.		2 nd Annual International SEMATECH Wafer Cleaning and Surface Preparation Workshop 2000, April 11-12, 2000, Hyatt Hotel, Austin, TX, 24 pages.	

Examiner Signature	Lynette J. Umez-Eronini	Date Considered	9/27/2004
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Examiner Name	Umez-Eronini, Lynette T.

Sheet

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2

Attorney Docket Number
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NON PATENT LITERATURE DOCUMENTS

Examiner Initials*	Cite No ¹	Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published	T ²
D.J.M-E MAR 8 2004 PATENT & TRADEMARK OFFICE		ANDEREGG, VON G., ET AL., Hydroxamatkomplexe III ¹). Eisen (III)-Austausch zwischen Sideraminen und Komplexonen Diskussion der Bildungskonstanten der Hydroxamatkomplexe, Helvetica Chimica Acta, Volumen XLVI, Fasciculus IV (1963) – No. 156, pgs. 1409-1422, Basel 7 (Schweiz).	
D.J.M-E		BIRUS, MLADEN, ET AL., Iron (III) Complexation by Desferrioxamine B in Acidic Aqueous Solutions. Kinetics and Mechanism of the Formation and Hydrolysis of the Binuclear Complex Diferrioxamine B, Inorganic Chemistry, Vol. 23, No. 14, 1984, pgs. 2170-2175, ©1984 American Chemistry Society.	
D.J.M-E		BIRUS, MLADEN, ET AL., Iron (III) Complexation by Desferrioxamine B in Acidic Aqueous Solutions. The Formation of Binuclear Complex Diferrioxamine B, Inorganica Chimica Acta, Vol. 78 (B6) N. 2, February 1983, pgs. 87-92, © Elsevier Sequoia/Printed in Switzerland.	
D.J.M-E		BIRUS, MLADEN, ET AL., Kinetics and Mechanism of Interactions Between Iron (III) and Desferrioxamine B. The Formation and Hydrolysis of Ferrioxamine B in Acidic Aqueous Solution, Croatica Chimica Acta, CCACAA 56 (1) pgs. 61-77.	
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D.J.M-E		BIRUS, MLADEN, ET AL., Mechanistic and Equilibrium Study of the Iron (III) Complexation by Deferriferrioxamine B in Aqueous Acidic Solution. Evidence for the Formation of Binuclear Diferrioxamine B, Inorganica Chimica Acta, Bioinorganic Chemistry Articles And Letters, Vol. 56(B3), No. 2, August 1981, pgs L43-L44, © Elsevier Sequoia S.A., Lausanne, Printed in Switzerland.	
D.J.M-E		EVERS, ANN, ET AL., Metal Ion Recognition in Ligands with Negatively Charged Oxygen Donor Groups. Complexation of Fe(III), Ga(III), In(III), Al(III), and Other Highly Charged Metal Ions, Inorganic Chemistry, Vol. 28, No. 11, 1989, pgs. 2189-2195, © 1989 American Chemical Society.	
D.J.M-E		GOULD, BRIAN, ET AL., A Thermodynamic Description of the Binding of Iron to Ferrioxamine B in Aqueous Solutions, Archives Of Biochemistry And Biophysics, Vol. 215, No. 1, April 1, 1982, pgs. 148-156, © 1982 by Academic Press, Inc., A Subsidiary of Harcourt Brace Jovanovich, Publishers, New York London Paris San Diego San Francisco São Paulo Sydney Tokyo Toronto.	
D.J.M-E		HARJU, LEO, The Stability Constants Of Some Metal Chelates Of Triethylenetetraminehexaacetic Acid (TTHA), Analytica Chimica Acta, Vol. 50, 1970, pgs. 475-489, Elsevier Publishing Company, Amsterdam, Printed in The Netherlands.	

Examiner Signature

Lynette T. Umez-Eronini

Date Considered
9/24/2004

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<i>L.J.N.E.</i> MAR 08 2004 U.S. PATENT & TRADEMARK OFFICE		HARJU, LEO, ET AL., Titrations With Complexing Agents Forming Mononuclear And Binuclear Complexes With Metals, <i>Analytica Chimica Acta</i> , Vol. 49, 1970, pgs. 205-219, Elsevier Publishing Company, Amsterdam, Printed in The Netherlands.	
<i>L.J.N.E.</i>		KHAN, M.M. TAQUI, ET AL., Aminopolycarboxylic Acid Complexes of Al(III), Ga(III) & In(III), <i>Indian Journal of Chemistry</i> , Vol. 19A, January 1980, pgs. 50-57, published by The Council Of Scientific & Industrial Research, New Delhi, India.	
<i>L.J.N.E.</i>		MA, RONG, ET AL., Protonation constants and metal ion binding constants of N,N'- bis(2-hydroxyphenyl)-N,N'-ethylenediaminediacetic acid, <i>Inorganica Chimica Acta</i> , The International Inorganic Chemistry Journal, Vol. 209, No. 1, July 1, 1993, pgs. 71-78, © 1993 Elsevier Sequoia.	
<i>L.J.N.E.</i>		MONZYK, BRUCE, ET AL., Kinetics and Mechanism of the Final Stage of Ferrioxamine B Aquation in Aqueous Acid, <i>Inorganica Chimica Acta</i> , <i>Bioinorganic Chemistry Articles and Letters</i> , Vol. 55 (B2) No. 1 January 1981, pgs. L5-L7, © Elsevier Sequoia S.A., Lausanne-Printed in Switzerland.	
<i>L.J.N.E.</i>		MONZYK, BRUCE, ET AL., Kinetics and Mechanism of the Stepwise Dissociation of Iron (III) from Ferrioxamine B in Aqueous Acid, <i>Journal Of The American Chemical Society</i> , Vol. 104, No. 18, 1982, pgs. 4921-4929, © 1982 American Chemical Society.	
<i>L.J.N.E.</i>		ÖHMAN, LARS-OLOF, Equilibrium and Structural Studies of Silicon (IV) and Aluminum (III) in Aqueous Solution. 21. A Potentiometric and ²⁷ Al NMR Study of the System H ⁺ -Al ³⁺ -MoO ₄ ²⁻ , <i>Inorganic Chemistry</i> , Vol. 28, No. 19, 1989, pgs. 3629-3632, © 1989 American Chemical Society.	
<i>L.J.N.E.</i>		WINSTON, ANTHONY, ET AL., Hydroxamic Acid Polymers. Effect of Structure on the Selective Chelation of Iron in Water, <i>Macromolecules</i> , Vol. 11, No. 3, May-June 1978, pgs. 597-603, © 1978 American Chemical Society.	
<i>L.J.N.E.</i>		YOSHIDA, ISAO, ET AL., New multidentate ligands. XXI. Synthesis, proton, and metal ion binding affinities of N,N',N"-tris[2-(N-hydroxycarbamoyl)ethyl]-1,3,5-benzenetricarboxamide (BAMTPH), <i>Canadian Journal of Chemistry</i> , Vol. 61, Number 12, December 1983, pgs. 2740-2744, National Research Council Canada, Printed in Canada by K.G. Campbell Corporation.	
<i>L.J.N.E.</i>		SCHWARZENBACH, VON G., ET AL., Hydroxamatkomplexe I. Die Stabilität der eisen (III)-Komplexe einfacher Hydroxamsäuren und des Ferrioxamins B, <i>Helvetica Chimica Acta</i> , Volumen XLVI, Fasciculus IV, No. 154, 1963, pgs. 1390-1400, Basel 7 (Schweiz).	
<i>L.J.N.E.</i>		VALTRON Specialty Chemicals for Tomorrow's Technology, VALTRON DP Series Formulated Detergents, 1 page.	

Examiner Signature	<i>Lynette T. Umez-Eronini</i>	Date Considered	<i>9/24/2004</i>
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